

## Preparation methods of nanoliposomes containing *Zataria multiflora* essential oil: A comparative study

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### ABSTRACT

Nowadays there is a substantial interest on the biological activities of essential oils. *Zataria multiflora* essential oil has a wide spectrum of pharmacological activities. However, essential oils are unstable and susceptible to degradation by oxygen, light and temperature and also low penetration in transdermal administration. Therefore, nanoliposomal systems can be used as promising strategies to overcome these limitations due to their unique set of properties: Nanoliposomes were prepared via two different methods based on thin film hydration method, including extrusion and sonication methods. The physical properties of nanoliposomes such as particle size, polydispersity index, zeta potential, encapsulation efficiency and stability were also determined. In comparison of two methods, sonicated nanoliposomes had smaller mean particle size and better dispersity while mean encapsulation efficiency in extruded ones were higher. It can be deduced by consideration of significance level; physicochemical properties of the vesicles were strongly influenced by essential oil and cholesterol concentration besides the preparation method.

**KEY WORDS:** NANOLIPOSOME; ZATARIAMULTIFLORA; ESSENTIAL OIL; THIN FILM HYDRATION; SONICATION

### INTRODUCTION

*Zataria multiflora* is one of the lamiacea family herbs with remarkable pharmacological properties. It natively grows in Iran, Pakistan, and Afghanistan (Saedi Dezaki

et al., 2016). It has various therapeutic effects such as antiseptic, antispasmodic, carminative, expectorant, anti-inflammatory, antiparasitic, spasmolytic, antiviral, antibacterial, antifungal and antioxidant properties (Sunar et al., 2009). Most of these properties are related to the main

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constituents of its essential oils, which are thymol and carvacrol and significant quantities of phenolic monoterpenes (Misaghi & Basti., 2007; Ali et al., 2000). Essential oils are confirmed to possess wide spectrum of pharmacological properties, besides their classical roles as natural food additives, such as the antioxidant, antibacterial, antifungal and anti-inflammatory activities (Shahsavari et al., 2008; Edris, 2007). Unfortunately, essential oils are biologically unstable, poorly soluble in water and are very sensitive to environment (Martin et al., 2007). All these obstacles restrict the application of essential oils as candidates for pharmatherapeutic treatments. Currently, nanoencapsulation of these oils in drug delivery systems, among which is the use of liposomal encapsulation, have been proposed to improve solubility, stability and efficacy of essential oil-based formulations (Belay et al., 2011; Saffari<sup>a</sup> et al., 2016).

Liposomes represent versatile and progressive nano delivery system for extensive range of biologically active compounds (Hofheinz et al., 2005). Liposome is a lipoidal vesicle composed of a bilayer membrane that have been developed to deliver drug to specific site in the body for more than four decades (Chetanachan et al., 2008); El-Samaligy et al., 2006). Because of biodegradation, non-toxicity, biocompatibility, non-immunogenicity, as well as superior efficacy, they are excellent carrier systems for a variety of applications, and in particular for essential oils (Sinico et al., 2005; Liolios et al., 2009; Saffari<sup>b</sup> et al., 2016). Encapsulation in liposome protect the essential oil from light, air and humidity, increases the solubility of the oil, enhancing the bioavailability of this drug, drug-targeting and makes the oil easier to handle (Wen et al., 2010; Valenti et al., 2001). The aim of this study was to prepare phosphatidyl choline based nanoliposomes via optimization the concentrations of essential oil, cholesterol and lipid composition through different methods for encapsulating zataria multiflora essential oil. Furthermore, to compare these methods via physicochemical properties of nanoliposomes including size, PDI index, zeta potential, encapsulation efficiency and drug release profile.

## MATERIALS AND METHODS

Lecithin, Egg PC, DOTAP were purchased from LIPOID (Germany). Cholesterol was from Sigma-Aldrich Co. The solvents (chloroform, methanol and ethanol) were purchased from commercial source and were in analytical grade. The essential oils were obtained from Barij essence Co. in Kashan, Iran.

### PREPARATION OF LIPOSOME

Liposomes can be formulated and processed to vary in size, composition and charge (Akbarzadeh et al., 2013).

For production of different types of liposomes, many methods are available (Meure et al., 2008). All Preparation methods can be simplified as to involve three basic steps: 1) dissolve the lipids in aqueous media in order to forming liposomes. 2) Purification of prepared liposomes. 3) Analysis of final product. Liposomes in our study prepared with three different methods as describe here:

#### 1. Thin film hydration

The most conventional and classical method, known by hydration of dried phospholipid films (Bangham et al., 1965). Multilamellar vesicles (MLVs) were prepared based on the thin film hydration method. Briefly the precise amount of lecithin, cholesterol and essential oil, according to several ratios of essential oil to total lipids, molar ratio of cholesterol to lecithin and lipid composition (DOTAP), were dissolved in chloroform-methanol (2:1, v/v) in a 500 ml round-bottomed flask. The organic solvent was eliminated by rotary evaporator (heidolph Hei-VAP Germany) under reduced pressure and high vacuum for 2 hour until a thin film was formed on the walls (above the lipid phase transition temperature, T<sub>c</sub>). Then the obtained lipid film was dispersed in Phosphate Buffer Saline solution (PBS, pH=7.4). This suspension was allowed to hydrate for 1.5 hour. Finally milky white suspension is formed. During the process, the conditions such as speed (120 rpm) and temperature (above the T<sub>c</sub> of lecithin) for conventional liposomes were maintained constant.

##### a. Concentration of essential oil

The total lipid concentration to prepare liposomes was 50 mM. Different ratios of essential oil to total lipid (1/2, 1/3, 1/4) with a constant molar ratio of cholesterol to phosphatidyl choline (1: 1) were prepared.

##### b. Concentration of cholesterol

After finding optimum concentration of essential oil to total lipid, cholesterol concentration was optimized by various concentrations as 7.1, 6.4, 4.83 mg/ml on a weight basis and 1:1, 1:2, 1:3 (molar ratio of Cholesterol: PC) in the overall formulation.

##### c. Charge-inducing lipids

EO-loaded cationic liposomes were formulated with pc, 1,2-dioleoyl-3-trimethylammonium-propane (DOTAP), and cholesterol at a 1:1:1 molar ratio of E80: DOTAP: CH.

### 2. Sonication method

Sonication is a simple method for reducing the size of liposomes to the nanoscale and manufacture of nanoliposomes. Liposomal suspension of MLVs transferred

Table 1. formulation based on method of preparation and composition

Method	Composition	EO /Total lipid	CH:PC
Thin film hydration	PC+CH+EO	0.5, 0.33,0.25	1:1
	PC+CH+EO	0.33	1:2,1:3
Sonication	PC+CH+EO	0.5, 0.33, 0.25	1:1
	PC+CHO+EO	0.33	1:2,1:3
Extrusion	PC+CH+EO	0.5, 0.33, 0.25	1:1
	PC+CH+EO	0.33	1:2,1:3

PC: phosphatidyl choline; CH: cholesterol; EO: essential oil

to a tube and placed in ice bath. Then the suspension was sonicated using prob sonicator (Hielscher UP400S, Germany).Vertically, the probe of a sonicator was fully immersed in the middle of the volume. The liposomal suspension was exposed to ultrasonic irradiation with an output of 70 Watt amplitude and duration of continuous 20 minute (6 times for 5 min). The energy input into liposomal dispersion is very high in this method.

**3. Extrusion method**

To obtain large unilamellar vesicles, the liposomal dispersion up to 3 mL was extruded three times through a polycarbonate membrane (100, 200 nm pore size) at above the lipid transition temperature by using an extruder to give a translucent solution.

Different formulation are listed in table 1 based on their method of preparation and their components and composition.

**Purification of Liposomes**

Entrapped vesicles were separated from non-entrapped vesicles using centrifugation method (Shashi et al., 2012). Vesicular dispersions were centrifuged at 10000 rpm for 10 minutes (laboratory centrifuge Hettich Universal 320 R). The supernatant was removed and the liposomes were reconstituted with PBS buffer. Concentration of essential oil in both fractions was determined.

**PHYSICOCHEMICAL CHARACTERIZATION OF LIPOSOMES**

**Particle Size distribution**

Mean vesicle size and polydispersity index (PDI) was determined by using Zetasizer (nanoZs Malvern ZEN 3600), which the principle is based on the Brownian motion of particles in medium. Dynamic light scattering is a simple and rapid method to determine the particle size and size distribution of liposomes.

**Zeta potential (ζ) determination**

Zeta potential is another important factor that is responsible for a description of the liposome surface charge and predicts the stability. Charge on loaded vesicles surface and average zeta potential was determined using Zetasizer ZEN 360 Malvern Instruments (Bhatia et al., 2004).

**Imaging of liposomes by Scanning Electron Microscopy (SEM)**

The shape and morphology of drug-loaded liposomal formulation were observed using a KYKY-EM3200 scanning electron microscopy (SEM, KYKY Instruments, China). Sample was dispersed on glass slide and gold paste used as filament and then viewed using an accelerating voltage of 20 kilovolt at different magnifications. Images of liposomes exhibited the diameter and size of vesicles and this is in agreement with the results of Zetasizer (Figure 4).

**Evaluation of the Encapsulation efficiency**

In case of purified liposomal suspension, after the centrifugation, by the sediment, the quantity of essential oil was measured using a UV Visible Spectrophotometer at λ = 270 nm. The encapsulation efficiency (EE %) of essential oil was calculated using the following formula:

$$\text{Encapsulation Efficiency} = \frac{S3/L3}{S2/L2} \times 100$$

Which in this equation; S3= EO amount in sediment & L3= phospholipid amount in sediment,

S2= EO amount before centrifuged & L2= phospholipid amount before centrifuged

**In vitro essential oil release studies**

Release studies were performed using dialysis membrane method. In vitro release was done for selected formula-

tion. In brief, 1000  $\mu$ L of the 7.1 mg/mL essential oil encapsulated liposome suspension was added in a dialysis bag (MWCO 12kDa, Thermo Fisher Scientific). The dialysis system was suspended in a release volume of 100 mL PBS at 25°C and rotated at 100 rpm (1:100 dilution relation between donor and acceptor compartments). At scheduled intervals, 1 ml of the release medium was collected for the UV- spectrophotometric assay. The same volume of fresh PBS buffer at the same temperature was added immediately to maintain constant release volume. The length of the dialysis tubing was kept consistent for all methods to ensure that the surface area available for dialysis remained constant. To ensure that dilution between the donor and acceptor compartments provided sink conditions, a 1:100 dilution study was conducted and release volume was set at 100 mL PBS pH 6.5.

## STATISTICAL ANALYSIS

Result has been reported as mean  $\pm$  standard deviation of three times repetition assays for each method. The mean values compared by T-Test and one-way analysis of variance (ANOVA), and statistical significance declared at  $P < 0.05$ .

## RESULTS & DISCUSSION

### MEAN PARTICLE SIZE

The results showed that the mean size of the vesicles was strongly affected by EO and CH concentration, besides of preparation method. Since the mean size of particles prepared by thin film method was around

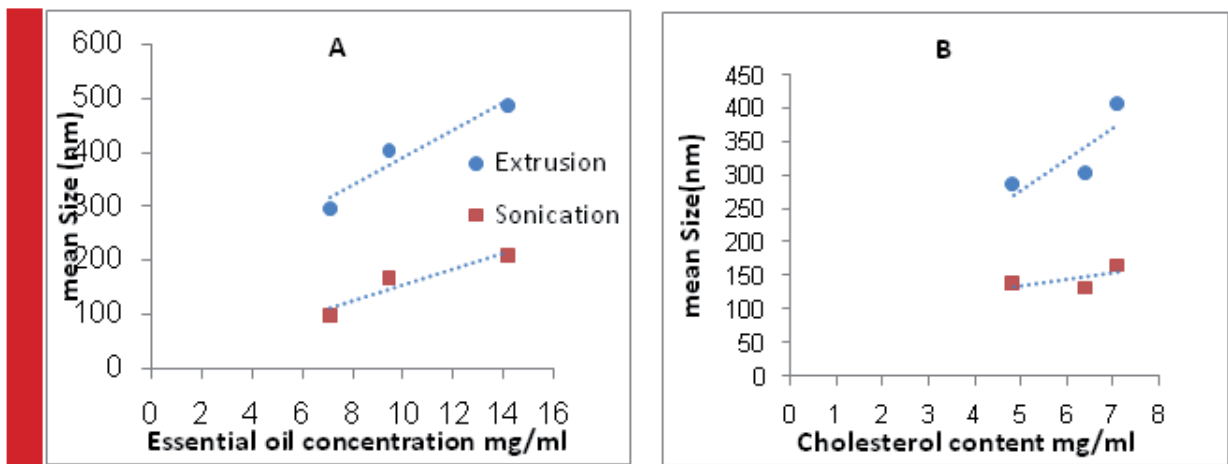


FIGURE 1. A is effect of essential oil concentration on Size of nanoliposomes and B is effect of cholesterol content on Size of nanoliposomes

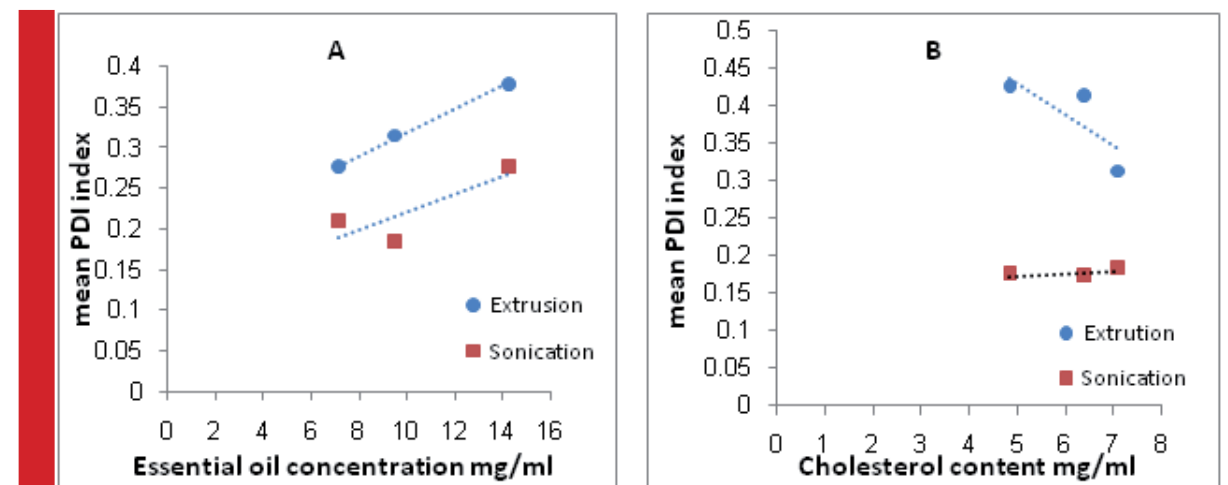


FIGURE 2. A is effect of essential oil concentration on PDI index and B is effect of cholesterol content on PDI index of nanoliposomes

Table 2. Effect of preparation parameters on Zataria multiflora essential oil Loaded nanoliposomes in two different preparation methods

	Essential oil concentration mg/ml	Size (nm)	PDI	EE%
Extrusion method	14.2	489.1±21	0.38±0.15	53.66±0.55
	9.46	405.66±67.5	0.31±0.03	56.26±1.99
	7.1	296.5±127.49	0.27±0.09	83.03±7.42
Sonication method	14.2	208.333±7.63	0.27±0.006	11.91±0.17
	9.46	167.333±2.08	0.18±0.007	14.13±0.36
	7.1	97.866±0.80	0.21±0.01	16.67±0.30

All data are mean value ± standard deviation

a few microns, 1426.3± 50 nm, particle size reduction in nano scale by two methods was quite impressive. According to the table 2, the mean size sonicated oil loaded vesicles ranged from about 97.8 to about 210 nm, while extruded liposomes were larger (256–489 nm). It's clearly seen that as the amount of the essential oil increases, size of vesicles increases. Also with decrease of cholesterol content, size of liposomes reduced (Figure 1).

**Polydispersity index**

All MLV liposomes prepared by thin film method had high PDI index (>0.4), which reflects the heterogeneity and dissimilarity of their colloidal system. While the size distribution of sonicated and extruded nanoliposomes was relatively narrow. Nanoliposomes prepared by the sonication method were most homogenous in size dis-

tribution than liposomes prepared by two other methods (Table 2).

As seen in figure 2 (A) by increasing the concentration of essential oil in two methods PDI was increased. In the case of cholesterol, figure 2 (B), in extruded method PDI decreases with increased cholesterol levels, this trend is more imperceptibly in sonication method.

**Zeta potential**

The zeta potential measurement of liposomal sample containing DOTAP was +14.7 mV. Other liposomes showed neutral charge that was owing to lack of charge ingredients.

**Encapsulation Efficiency**

The encapsulation efficiency greatly depends on liposomal content, lipid concentration, method of prepara-

Table 3. Effect of cholesterol concentrations on properties of Zataria multiflora essential oil loaded nanoliposomes in two different preparation methods

	Cholesterol: lecithin	Size(nm)	PDI	EE%
Extrusion method	1:1 (7.1mg/ml)	405.66±67.50	0.313±0.03	56.26±1.99
	1:2 (6.4mg/ml)	302.66±62.54	0.412±0.03	62.39±4.50
	1:3 (4.83mg/ml)	285.033±15.59	0.424±0.12	70.56±2.58
Sonication method	1:1	167.33±2.08	0.183±0.01	14.16±0.31
	1:2	133.66±0.57	0.172±0.001	18.59±0.01
	1:3	138±1	0.174±0.0005	18.96±0.05

All data are mean value ± standard deviation

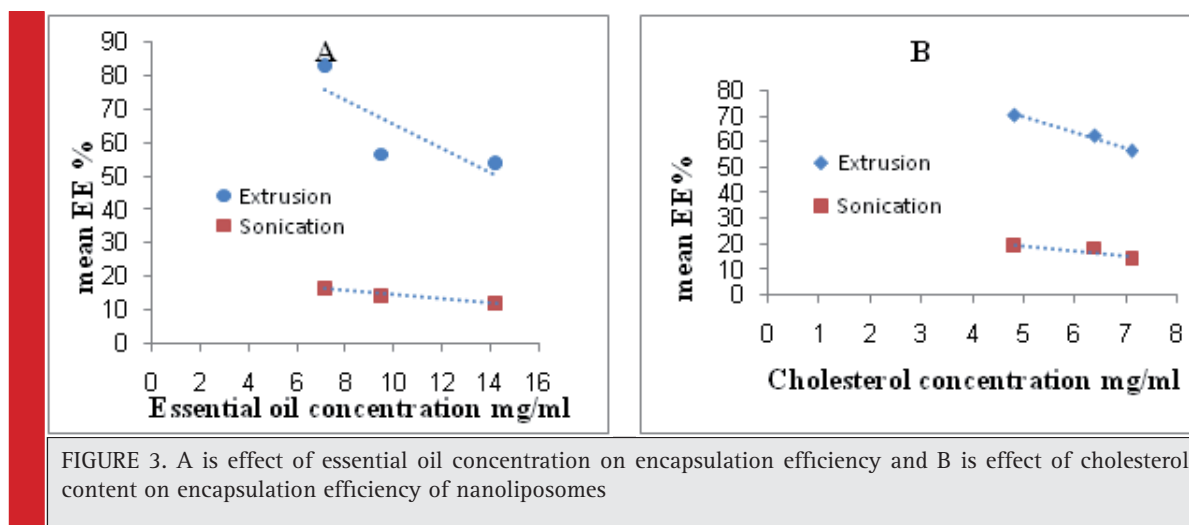


FIGURE 3. A is effect of essential oil concentration on encapsulation efficiency and B is effect of cholesterol content on encapsulation efficiency of nanoliposomes

tion and the drug which is used. The results show that both two methods have acceptable incorporation efficiency. However sonication method gave a lower content in incorporated essential oil than extrude method (11-18%). Extruded vesicles showed high encapsulation efficiency (53-83%) of the essential oil. The highest mean entrapment efficiency was found at 0.25 of EO/total lipid (Table 2). Moreover, in the case of cholesterol effect, it was also observed that liposomal formulation PC/Chol (3:1) load the highest amount of the active ingredient in each method ( $70.56 \pm 2.58$  % for extruded vesicles and  $18.96 \pm 0.05$  % for sonication method) (Table 3). From figure 3(A), we observed a decrease of incorporated oil with increase of essential oil concentration in each method. Also, it can be appreciated that decrease content of cholesterol lead to increase loading of the

essential oil (figure 3 B). Results were summarized in Table 2 and 3. On the whole, ANOVA test showed that oil concentration influence on physicochemical properties of each method significantly ( $p < 0.05$ ).

From figure 3(A), we observed a decrease of incorporated oil with increase of essential oil concentration in each method. Also it can be appreciated that decrease content of cholesterol lead to increase loading of the essential oil (figure 3B).

#### Image of selected nanoliposome

SEM image of prepared nanoliposomes has been shown in Figure 4. As it can be seen all nanoliposomal particles have spherical structure and system is homogenous in both methods. Both formulations in which used EO: PC: CH 1:4:1. Liposomes prepared by sonication method

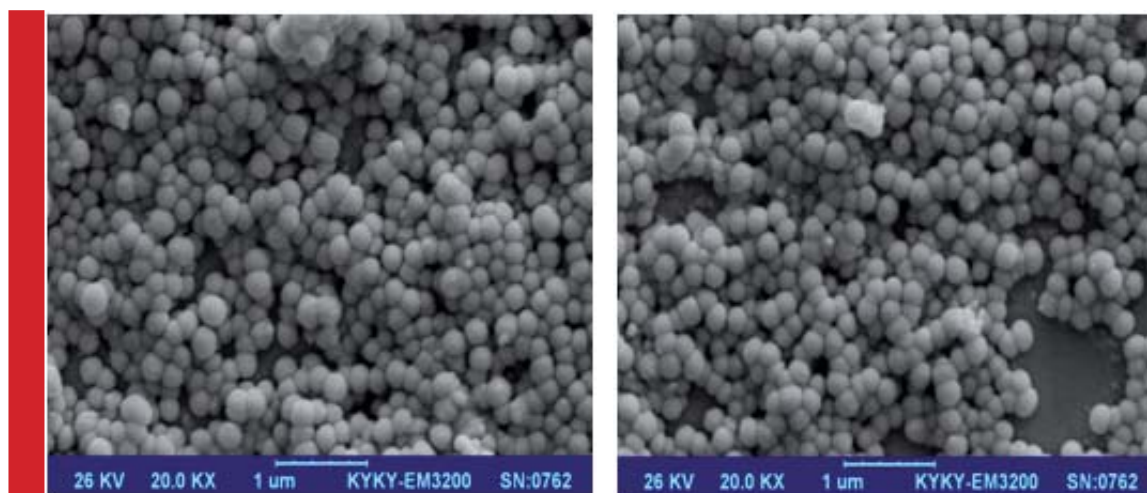


FIGURE 4. SEM image of Zatariamultifloraessential oil loaded liposomes by sonication (right image) and extrusion (left image) method.

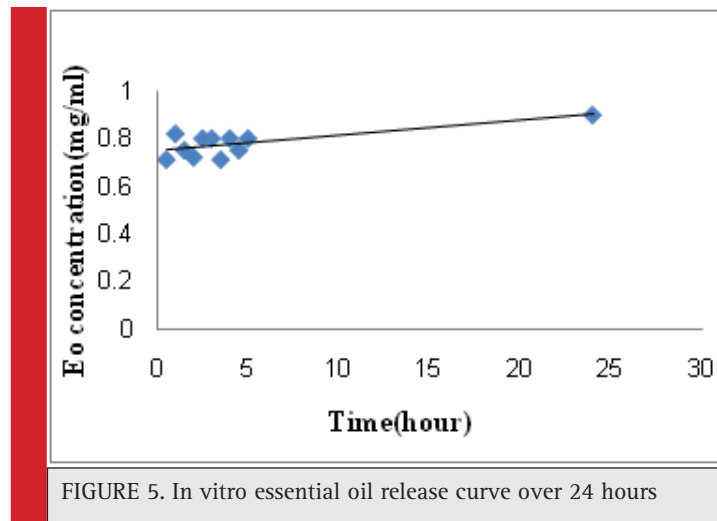


FIGURE 5. In vitro essential oil release curve over 24 hours

(right image) were smaller than liposomes treated via extrusion method (left image). It can be due to higher content of essential oil that fluidize liposomal bilayers and increase their susceptibility for aggregation.

#### In vitro essential oil release studies

As showed in Figure 5, the release of EO within 24 hours from nanoliposomes was very inconsiderable and negligible. In the other words, it can be said that EO leakage from liposomes does not occur. The selected formulation was EO: PC: CH 1:4:3 nanoliposomes containing Zataria essential oil.

#### Liposome stability

The physical and chemical stability of sonicated formula were evaluated at 5°C, 25°C and 37°C for two months. The sonicated formula (1/4) loaded with Zataria essential oil were stable for at least 8 weeks at room and refrigerator temperatures as the particle size and the EE% of the liposomes did not change significantly during this period (data not shown).

#### Effect of essential oil concentration on liposomes characteristics

Different methods presented different results in the formation of the nano-sized liposomes. Due to high energy used in probe sonicator, liposomes prepared by this method have, smaller mean size and PDI index compared to extruded ones. Sonication generally produces SUV liposomes with lower size and lower incorporated essential oil than MLVs that are according to Sinico et al findings (Sinico et al., 2005). Comparison the average size of nanoliposomes in different concentration of essential oil in two methods showed significant difference ( $p < 0.05$ ). It has been showed that nanoliposomes size could be affected by EO amount incorporated.

Also entrapment of lipophilic compounds in liposome membrane depends on size of liposome (Schwendener & Schott, 2017). Varona et al, also showed an essential oil amount impact on the liposome size (Varona et al., 2011). According to our results, by increasing the concentration of essential oil to total lipid with the same ratio of PC: CH, average size of the nanoparticles have increased. Probably because of higher EO concentration required greater mechanical force in the reduction size process of vesicles. In study of Detoni and colleagues, it was not observed any changes in the size of the nanoparticles containing different ratios of *Zanthoxylum tin-goassubia* oil to the phospholipid, which is inconsistent with our study (Detoni et al., 2009).

PDI (poly dispersity index) usually is considered to be as an indicator of colloidal particle size distribution. The smaller index indicates the particle distribution is more limited and so the system will be homogeneous and more uniform in size (Ruozi et al., 2005). Two methods had low value PDI index although the mean PDI index in sonication method is lower than the extrusion method. In both methods, the mean PDI index changes made no significant difference, in different concentration of EO ( $0.05 < P$ ). The liposomes encapsulated *atractylodes macrocephala* essential oil that formed by the modified RESS had uniform and narrow size distribution (Wen et al., 2010). The same results have been obtained by Valenti et al and Sinico et al which is consist with our study (Sinico et al., 2005; Valenti et al., 2001). On the contrary, Yoshida et al reported PDI index higher than 0.7 for their MLV dispersions using *Eugenia uniflora* EO (Yoshida et al., 2010). According to the above figure (3), it can be observed that, in both two methods, the incorporation efficiency decreased when the EO amount was increased. Ortan et al, showed a slight decrease in the encapsulation efficiency of *anethum graveolens* essential

oil with the increase of its concentration which is consistent with present study (Ortan *et al.*, 2009). Lavandin and his colleagues obtain varying amounts of encapsulation efficiency by loading different ratio of essential oil to phospholipid (Varona *et al.*, 2011). With increasing of essential oil concentration the encapsulation efficiency increases because of more EO can be encapsulated into liposomes that is consistent by the studies of Fang *et al* and also Varona *et al* which in contrast with present study (Varona *et al.*, 2011; Fang *et al.*, 2001). The difference in the encapsulation efficiency was explained by Detoni *et al.* According to their study, the physical and chemical properties of essential oils, or changes may be created in the techniques of liposomes preparation such as concentration of phospholipids, cholesterol and the ratio of EO to total lipid can affect this property (Detoni *et al.*, 2012). In our study, Significance differences in incorporation efficiency related to the preparation method was observed ( $p < 0.05$ ). It has been reported that produced liposomes by sonication method, may be have lower encapsulation efficiency due to the degradation possibility of phospholipids and encapsulated compounds (Riaz, 1996). In a similar study, Zataria multiflora essential oil into the nanoliposomes was encapsulated. The essential oil encapsulation within liposomes were 18 to 22 % by providing multiple ways (Khatibi *et al.*, 2015). Celia *et al.*, have incorporated bergamot essential oil by using extrude method with an encapsulation efficiency of 75% (Ceilia *et al.*, 2015). The result obtained in present study reconfirms approximately their findings.

#### Effect of cholesterol content on liposomes characteristics

By maintaining the ratio of EO to total lipid (1/3) in the formulations, the effect of different cholesterol concentration on liposome features was also determined. Cholesterol as a usual composition plays an important role in the structure of the liposome. Cholesterol due to its amphiphilic properties, perch into liposome so that a hydroxyl group to the core of water oriented and the hydrophobic tail to the phospholipid bilayer of the liposomes. In previous studies it has been widely reported that the formation and stability of the liposomes is highly dependent on the ratio of cholesterol to phospholipid, so have great impact on the behavior of the carrier in vitro and in vivo conditions (Haeri *et al.*, 2014). Other effects of cholesterol are control and reduce membrane permeability, making of hardness to membrane and layer structural stability (Chan *et al.*, 2004). Results shown in figure 1 part B demonstrated that liposome size increased with the increasing amount of cholesterol in both two methods. Varona and colleagues observed that with lower cholesterol levels, the size of the liposomes reduced which it's in line with the

results of this study (Varona *et al.*, 2011). It is reported that the insertion of cholesterol in liposomal membrane causes liposomes more rigid and resistant to size reduction (Briuglia *et al.*, 2015). Comparison the mean size of two methods, sonication method showed a significant size reduction ( $p < 0.05$ ). Also the mean PDI index in different molar concentrations of cholesterol in two methods showed that changes are significant ( $p < 0.05$ ). Sonication method gave smaller and more homogeneous nanoliposomes than extrude method. Lipophilic molecules, competes with cholesterol molecules for the lipophilic space in the lipid bilayer (Jaafar-Maalej *et al.*, 2010) so, cholesterol might distaste the incorporation of hydrophobic molecules into the liposome bilayer membrane (Fathi Moghaddam *et al.*, 2008; Rezaee *et al.*, 2015; Saffari *et al.*, 2013). Ortan *et al* in their study showed that by increasing the amount of Cholesterol, amount of encapsulated EO (EE%) in liposomes reduced (Ortan *et al.*, 2009). This result was confirmed by Varona *et al* via essential oil of lavandi (Varona *et al.*, 2011). The same result was obtained in present study with both two methods and encapsulation efficiency in both methods have significantly decreased ( $p < 0.05$ ).

In summary, this study contributes to the understanding effect of preparation method & the different concentration of EO, Cholesterol content on Zataria essential oil loading into liposomes. The study presented here suggested that the physicochemical features of loaded nanoliposomes clearly influenced by EO concentration, cholesterol content and liposome preparation method. It can be deduced by consideration of significance level, sonication method gave smaller nanoliposomes than the extrude method, while higher EO incorporation obtained by extruded method.

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