

COD reduction from refinery wastewater using SiO₂ photocatalyst synthesized by wheat husk

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ABSTRACT

In this research work wheat husk has been used as silica source for synthesis of SiO₂ nanocatalyst by sol-gel method. High concentration of COD is found in refinery wastewater as pollutant. The size of synthesized SiO₂ nanoparticles found from 5 to 30 nm range. AFM, FEG-SEM and TEM analysis confirmed that synthesized nanocatalyst are in nano range. FTIR, XRD, and EDAX analysis confirmed that catalyst is SiO₂. The COD was removed by photocatalytic reaction in aluminium UV-lamp photoreactor. The optimum percent removal of COD is found to be 85% at 8 hour reaction time and 250C temperature and at 9.3 pH. BET surface area is found to be 300m²/g.

KEY WORDS: SiO₂, NANOCATALYST, WHEAT HUSK, COD, REFINERY WASTEWATER.

INTRODUCTION

Wheat husk contain silica in abundant quantity. Wheat husk has been used as agriculture waste to synthesis SiO₂ nanocatalyst. Refinery wastewater contains COD, BOD, Total hydrocarbon (TOH), and phenolic compound as pollutants (Uddeen et al., 2011, Yu et al., 2016). Industrial wastewater is big challenge for environment (Oubrayame et al., 2015). Wheat husk has been used as

agriculture waste for synthesis of SiO₂ so it is green synthesis method.

In this research work COD has been treated from refinery waster and effect of various parameters like pH, and time have been analyzed. In this work UV-lam aluminum photocatalyst has been used for 8 hour reaction time at 6 h optimum percent removal has been found. By aluminum photocatalytic reactor intensity of light has been controlled to get maximum percent removal.

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Various characterization like FEG-SEM, TEM, XRD, FTIR, AFM have been carried out.

MATERIALS AND METHODS

MATERIALS

All chemicals used in this experiment were used to analytical grade (AR). Orthophosphoric acid (85% purity) acid to make 1 M solution, NaOH to make a 1N solution to maintained pH, double distilled water, refinery wastewater from Northern refinery industry.

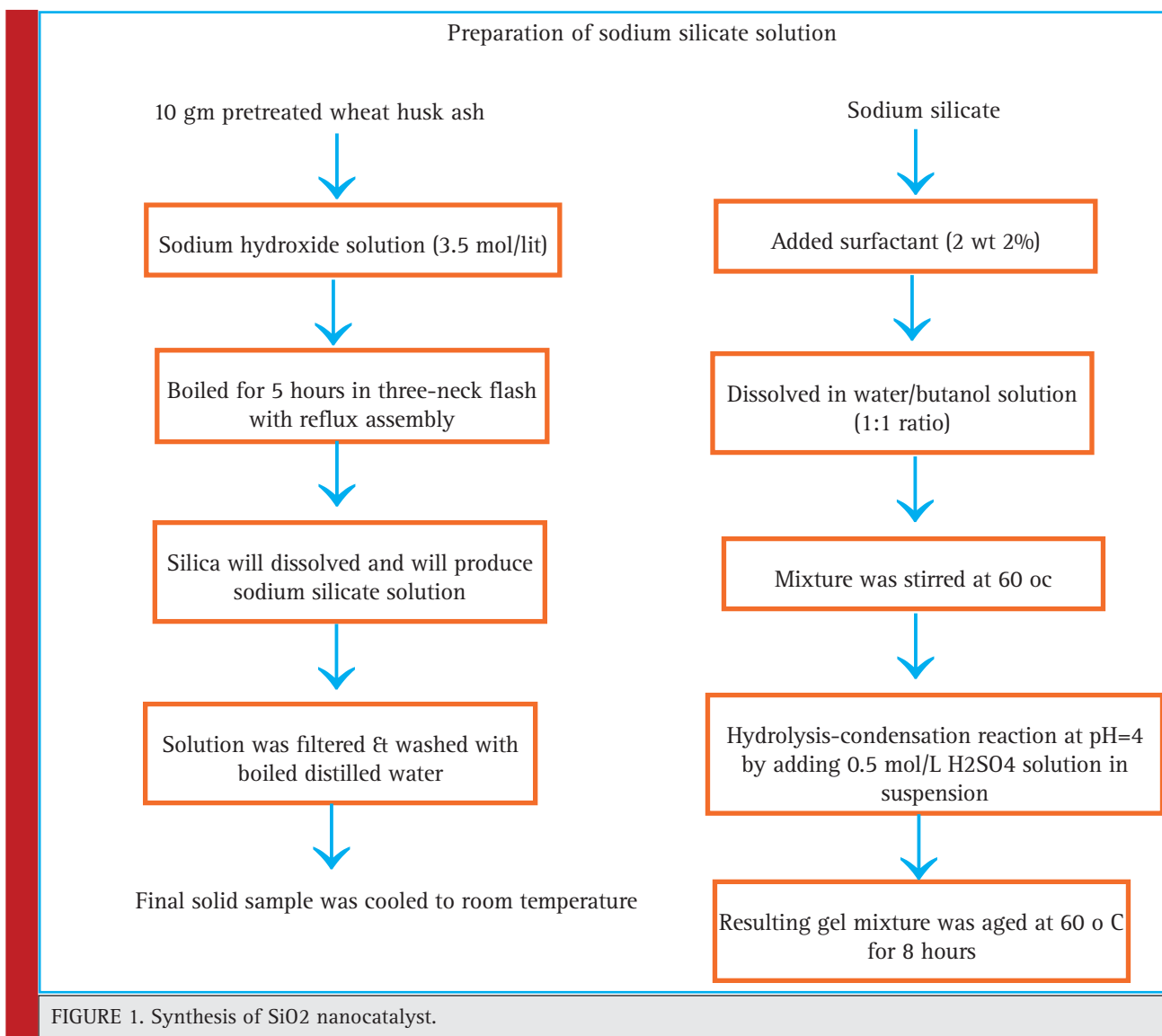
METHODS

Synthesis of SiO₂ nanocatalyst by sol-gel method

Wheat husk was collected as agriculture waste from agriculture land, then it was washed and dried in oven at 60°C. Silica was extracted from wheat husk and nanocatalyst of SiO₂ was made by sol-gel method the procedure has been shown in flow diagram 1 (Gayen et al., 2011). 10 gm wheat husk was activated by orthophosphoric acid.

Characterization

FEG-SEM, TEM, XRD, FTIR, AFM, EDAX had been carried out to know size, shape and morphology of nanocatalyst.



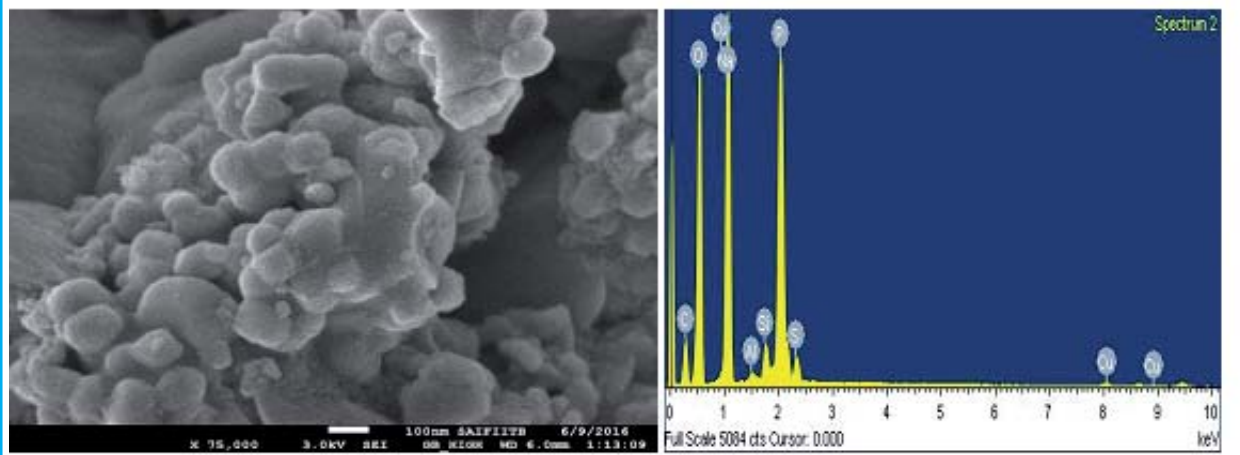


FIGURE 2. Shows FEG-SEM Image EDX Image.

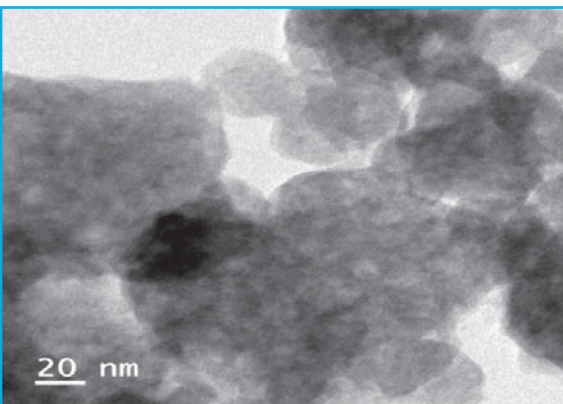


FIGURE 3. TEM Image.

FEG-SEM, TEM and AFM confirmed that particles are in nano range. XRD confirmed phase identification of SiO_2 nanocatalyst (Mourhly et al., 2015). FTIR test shows Si-O-Si stretching for nanocatalyst and for other functional groups present. TEM and FEG-SEM FTIR were carried out at SAIF, IIT Mumbai. AFM test was carried out at North Maharashtra university, Jalgaon. AFM was performed at MANI, Bhopal.

Treatment method

For treatment of refinery wastewater aluminum photo-catalytic reactor was used. The reaction was carried out in three neck flask. The light source was taken UV-lamp. There are Six UV-lamp are there in aluminum reactor. By this arrangement intensity of light can change.

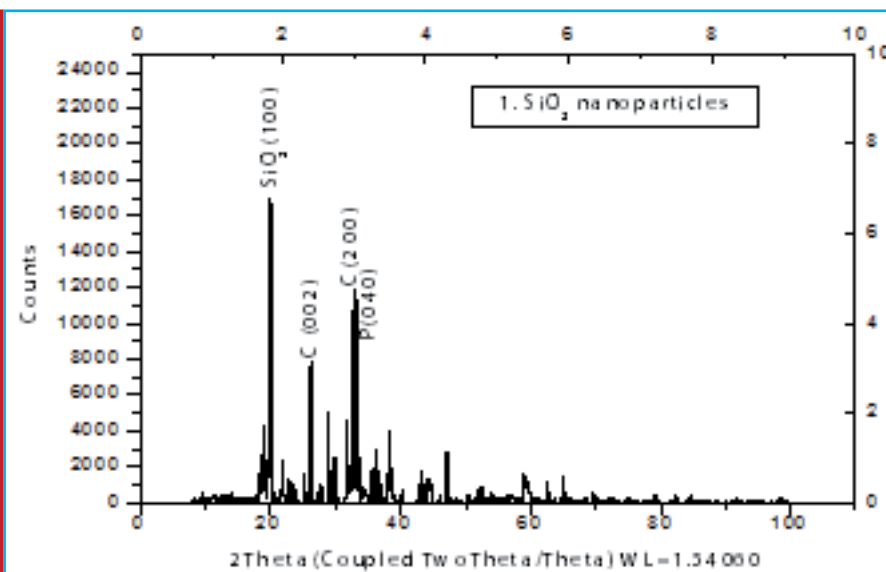


FIGURE 4. XRD analysis.

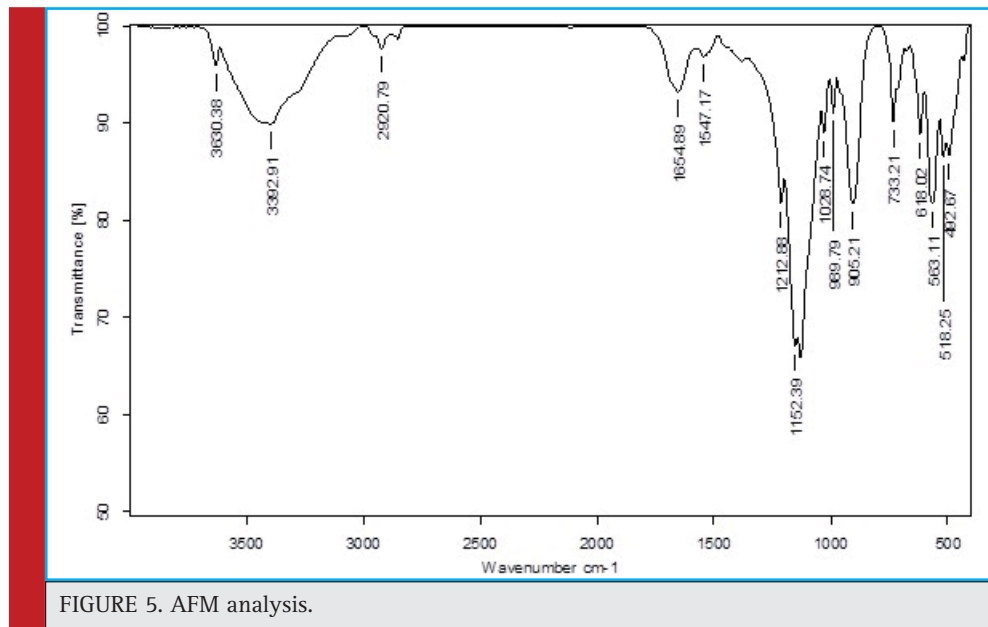


FIGURE 5. AFM analysis.

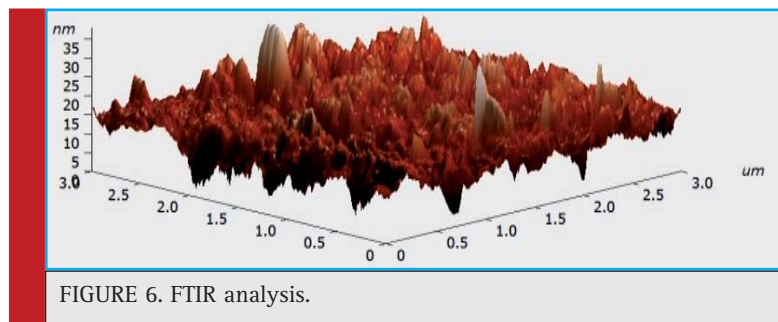


FIGURE 6. FTIR analysis.

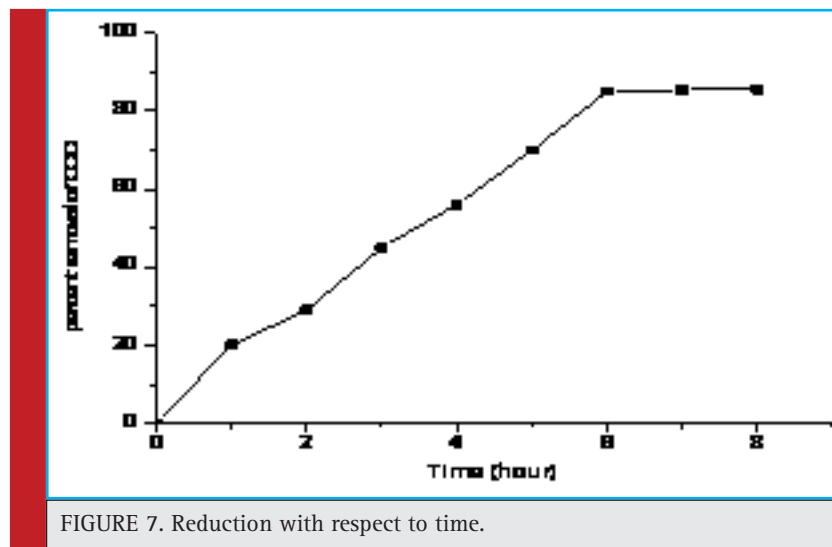


FIGURE 7. Reduction with respect to time.

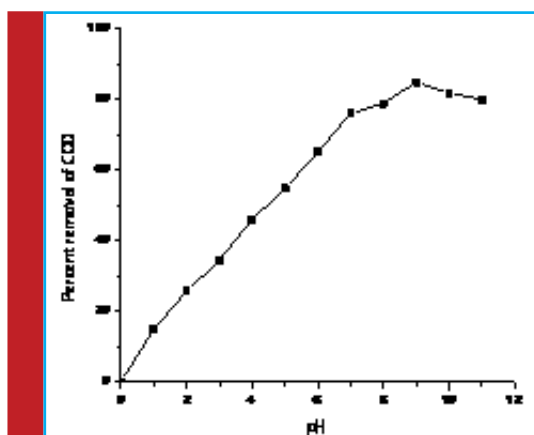


FIGURE 8. COD reduction with respect to pH.

The photocatalytic reaction was performed for 8 hours. The samples were analyzed at different time intervals 1h, 2h, 3h, 4h, 5h, 6h, 7h, 8h. The reaction was performed at different catalyst loads 0.5g/L, 1.0g/L and 1.5g/L. The optimum percentage removal is 85% at 0.5g/L catalyst load for refinery wastewater and 9.3 pH.

RESULTS AND DISCUSSION

Characterization: Results shown by characterization found that synthesized catalysts are in nano range.

Treatment of refinery wastewater:

Synthesized nanocatalyst by wheat husk shows good results to treat refinery wastewater by photocatalytic reaction (Choquette et al., 2014).

CONCLUSION

From characterization of nanocatalyst it has been concluded that synthesized nanocatalyst are irregular in shape. The size of nanocatalyst is from 5 to 25

nm. Synthesized nanocatalyst are crystalline in nature. The optimum percent removal of COD from refinery wastewater is 85% at 6 hour reaction time. The optimum pH for COD removal is 9.0. The results show that synthesized nanocatalysts are good photocatalyst for COD removal from refinery wastewater.

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